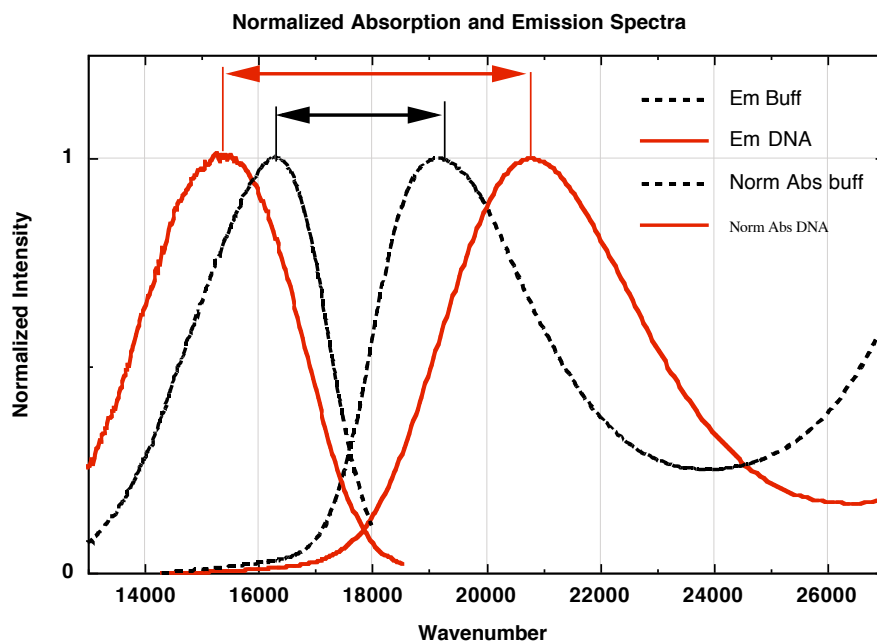


CHM 598, Photochemistry, Spring 2005, Homework 4, my answers!

$$\text{Stokes Shift (DNA)} = 20790 - 15400 \text{ cm}^{-1} = 5390 \text{ cm}^{-1} = 0.67 \text{ eV} = 15.4 \text{ kcal/mol}$$

$$\text{Stokes Shift (Buff)} = 19130 - 16290 \text{ cm}^{-1} = 2840 \text{ cm}^{-1} = 0.35 \text{ eV} = 8.1 \text{ kcal/mol}$$

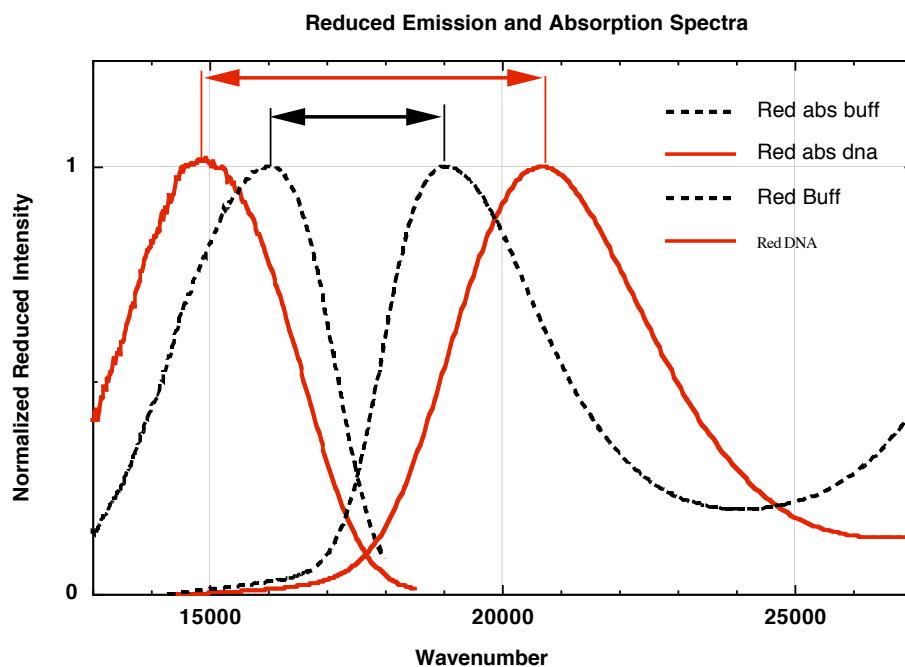


$$\text{Total } \lambda \text{ (DNA)} = 20700 - 14850 \text{ cm}^{-1} = 5850 \text{ cm}^{-1} = 0.73 \text{ eV} = 16.7 \text{ kcal/mol}$$

$$\Delta E_{0,0} \text{ (DNA)} = 17680 \text{ cm}^{-1} = 2.19 \text{ eV} = 50.6 \text{ kcal/mol}$$

$$\text{Total } \lambda \text{ (Buff)} = 19050 - 16050 \text{ cm}^{-1} = 3000 \text{ cm}^{-1} = 0.37 \text{ eV} = 8.6 \text{ kcal/mol}$$

$$\Delta E_{0,0} \text{ (Buff)} = 17520 \text{ cm}^{-1} = 2.17 \text{ eV} = 50.1 \text{ kcal/mol}$$



Determination of average emission frequency:

Evaluation of $\int I_f(\tilde{\nu}) d\tilde{\nu}$, for the normalized to unity emission spectrum in DNA,
from 18519 to 12516 $\text{cm}^{-1} = 3305.8$

Evaluation of $\int I_f(\tilde{\nu}) \tilde{\nu}^{-3} d\tilde{\nu}$, for the normalized to unity emission spectrum in DNA,
from 18519 to 12516 $\text{cm}^{-1} = 9.65 \times 10^{-10}$

average emission frequency in DNA = $(3305.8/9.65 \times 10^{-10})^{-3} = 15075 \text{ cm}^{-1}$
(this value really should be smaller, but we don't have the entire spectrum)

Evaluation of $\int I_f(\tilde{\nu}) d\tilde{\nu}$, for the normalized to unity emission spectrum in Buffer,
from 18519 to 12516 $\text{cm}^{-1} = 2811.9$

Evaluation of $\int I_f(\tilde{\nu}) \tilde{\nu}^{-3} d\tilde{\nu}$, for the normalized to unity emission spectrum in Buffer,
from 18519 to 12516 $\text{cm}^{-1} = 7.33 \times 10^{-10}$

average emission frequency in Buffer = $(2811.9/7.33 \times 10^{-10})^{-3} = 15654 \text{ cm}^{-1}$

Determination of radiative rate:

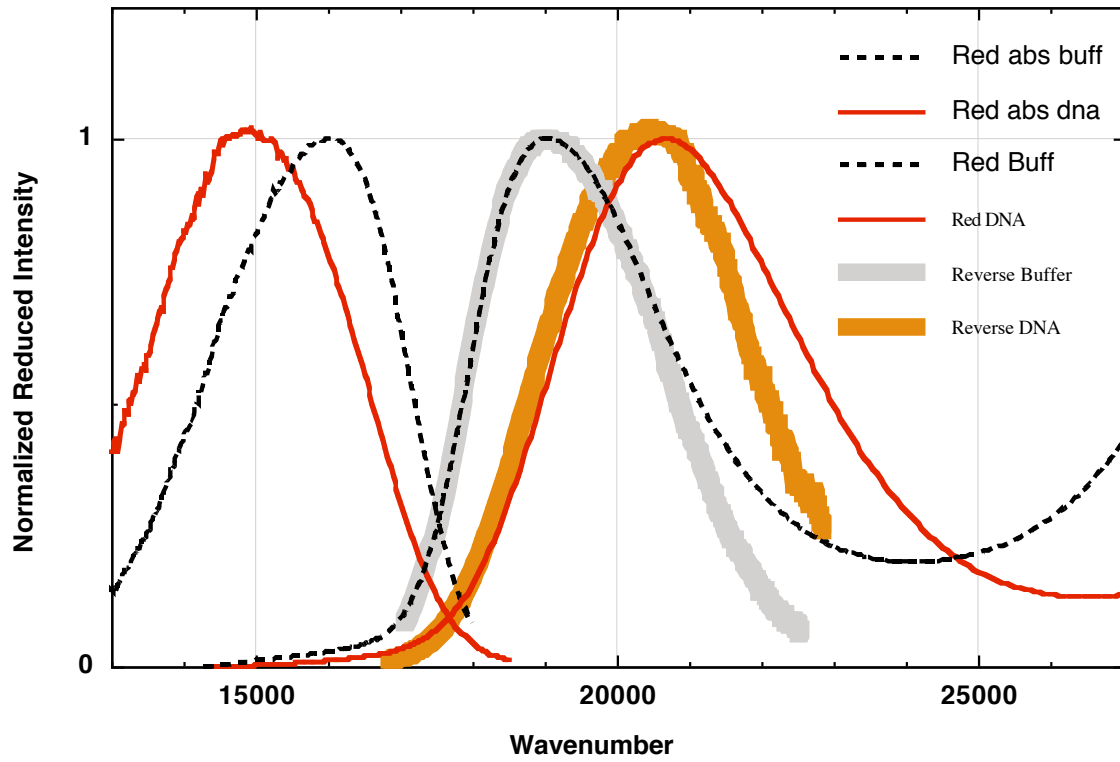
Evaluation of $\int \frac{\epsilon_{abs}(\tilde{\nu})}{\tilde{\nu}} d\tilde{\nu}$ from 14000 to 23500 cm^{-1} in Buffer = 1090.6

$$k_f = 2.881 \times 10^{-9} (1.335)^2 (15654)^3 1090.6 = \mathbf{2.15 \times 10^7 \text{ s}^{-1}}$$

Evaluation of $\int \frac{\epsilon_{abs}(\tilde{\nu})}{\tilde{\nu}} d\tilde{\nu}$ from 14000 to 26500 cm^{-1} in DNA = 779.7

$$k_f = 2.881 \times 10^{-9} (1.335)^2 (15075)^3 779.7 = \mathbf{1.37 \times 10^7 \text{ s}^{-1}}$$

Reduced Emission and Absorption Spectra and Reversed Emission Spectra



These plots show that the emission spectrum in buffer exhibits an good mirror image relationship with the absorption spectrum, whereas that in DNA does not